

The silver and sodium ions are of approximately equal size (the radii are 0.95 and 1.13×10^{-10} m, respectively⁴⁵), but the compressibilities of the silver compounds are only about one-half those of the sodium salts at the same temperature (see table 5). The differences between the ΔV_A values are therefore consistent with the arguments advanced above. However, there is evidence that the fused silver halides do not have 100 % ionic character,^{46, 47} and probably the covalent contribution to the bonding is pressure dependent.⁴⁸ The application of the Rice-Allnatt theory in these circumstances is open to objection. Similar considerations would apply to the fused thallous halides.⁴⁹

We are grateful to Dr L. V. Woodcock, S. Sternberg and V. Vasilescu for allowing us to use their unpublished results, to Dr. H. Bronstein for advice on the fabrication of sapphire conductivity cells and to the S.R.C. for an equipment grant and Fellowship (P. N. S.).

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